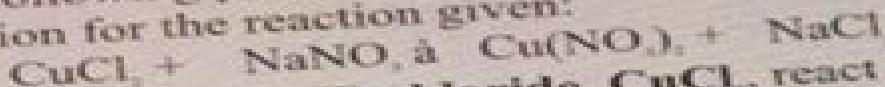


[Limiting Reactant and Percent Yield
Worksheet]

Answer the following problems with complete solutions.

1. The equation for the reaction given:

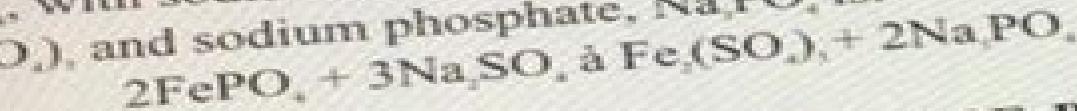


a) If 0.11 mol of copper (II) chloride, CuCl_2 , react with 0.24 mol of sodium nitrate, NaNO_3 , how many moles of sodium chloride, NaCl , can be formed?

b) What is the name of the limiting reagent?

c) If 11.3 grams of sodium chloride (actual yield) are formed in the reaction, what is the percent yield of this reaction?

2. The balanced equation for the reaction of iron (III) phosphate, FePO_4 , with sodium sulfate, Na_2SO_4 , to make iron (III) sulfate, $\text{Fe}_2(\text{SO}_4)_3$, and sodium phosphate, Na_3PO_4 , is:



a) If you perform this reaction with 0.17 mol of FePO_4 and an excess of Na_2SO_4 , how many moles of $\text{Fe}_2(\text{SO}_4)_3$ can you make?

b) If 18.5 grams of $\text{Fe}_2(\text{SO}_4)_3$ are actually made when you do this reaction, what is your percent yield?

Limiting Reactant And Percent Yield Lab Answers

Steven S. Zumdahl, Paul B. Kelter

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approach me from past years and ask about the forensics case and specific characters from the story I have never had a student come back to me and comment on that unit with the multiple choice test at the end from the Introduction to Forensics in Chemistry The Murder of Kirsten K How did Kirsten K's body wind up at the bottom of a lake and what do wedding cake ingredients soil samples radioactive decay bone age blood stains bullet matching and drug lab evidence reveal about whodunit These mysteries are at the core of this teacher resource book which meets the unique needs of high school chemistry classes in a highly memorable way The book makes forensic evidence the foundation of a series of eight hands on week long labs As you weave the labs throughout the year and students solve the case the narrative provides vivid lessons in why chemistry concepts are relevant and how they connect All chapters include case information specific to each performance assessment and highlight the related national standards and chemistry content Chapters provide Teacher guides to help you set up Student performance assessments A suspect file to introduce the characters and new information about their relationships to the case Samples of student work that has been previously assessed and that serves as an answer key for you Grading rubrics Using Forensics in Chemistry as your guide you will gain the confidence to use inquiry based strategies and performance based assessments with a complex chemistry curriculum Your students may gain an interest in chemistry that rivals their fascination with Bones and CSI Operational Organic Chemistry John W. Lehman,1988

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