

**Reminders:**

1. In a neutral atom the number of protons equals the number of electrons.
2. An atom can NEVER gain or lose protons.
3. The number of protons equals the atomic number

# Ion Practice Set

**1. What is an ion?**

Es un átomo o un grupo de átomos (molécula) con una carga negativa o positiva

**2. What does the number next to the ions signify?**

La carga que llevan

**Complete the following table, using the periodic table in the back of your book.**

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED
ex	Fluorine	F <sup>-</sup>	9	10	gained one
1	Yodo	I <sup>-</sup>	53	54	Gained one
2	Azufre	S <sup>-2</sup>	16	18	gained two
3	potassium	K <sup>+</sup>	19	18	lost one
4	Calcio	Ca <sup>+2</sup>	20	18	Lost two
5	Bromo	Br <sup>-</sup>	35	36	Gained one
6	Estroncio	Sr <sup>+2</sup>	38	36	Lost two
7	Hidrogeno	H <sup>+</sup>	1	2	Gained one
8	Oxigeno	O <sup>-2</sup>	8	10	gained two
9	Magnesio	Mg <sup>+2</sup>	12	10	lost two
10	aluminum	Al <sup>+3</sup>	13	10	Lost two
11	Selenio	Se <sup>-2</sup>	34	36	Gained two
12	Hidrogeno	H <sup>+</sup>	1	0	Lost one
13	lithium	Li <sup>+</sup>	3	2	lost one
14	Rubidio	Rb <sup>-</sup>	37	38	Gained one
15	Cloro	Cl <sup>-</sup>	17	18	Gained one

# Ion Practice Set Answer Key

**Michigan. State Board of Agriculture**

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