





- (a) 
- (b) 
- (c) 
- (d) 

**Solution:**

Option (iv) is the answer.

5. Which of the following is true about the value of the refractive index of quartz glass?

- (i) Same in all directions
- (ii) Different in different directions
- (iii) Cannot be measured
- (iv) Always zero

**Solution:**

Option (i) is the answer

6. Which of the following statement is not true about amorphous solids?

- (i) On heating, they may become crystalline at a certain temperature.
- (ii) They may become crystalline on keeping for a long time.
- (iii) Amorphous solids can be moulded by heating.
- (iv) They are anisotropic.

**Solution:**

Option (iv) is the answer.

7. The sharp melting point of crystalline solids is due to \_\_\_\_\_.

- (i) a regular arrangement of constituent particles observed over a short distance in the crystal lattice.
- (ii) a regular arrangement of constituent particles observed over a long distance in the crystal lattice.
- (iii) the same arrangement of constituent particles in different directions.
- (iv) a different arrangement of constituent particles in different directions.

**Solution:**

Option (ii) is the answer.

# Mastering Chemistry Chapter 7 Answers

**James C. Hill**



## Mastering Chemistry Chapter 7 Answers:

**A Problem-Solving Approach to Aquatic Chemistry** James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Mastering Chemistry Chapter 7 Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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