

# Limiting Reactant Worksheet

- 1 Balance the following reaction:



- a Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

- b How many grams of excess reactant are left?

- 2 Acrylonitrile ( $\text{C}_3\text{H}_3\text{N}$ ) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene ( $\text{C}_3\text{H}_6$ ) by reacting with nitric oxide (NO) as follows:



What mass of  $\text{C}_3\text{H}_3\text{N}$  can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

# Limiting Reactant Practice Sheet Key Unit 8

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A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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Introductory Chemistry Spencer L. Seager, Michael R. Slabaugh, 1979 Chemical Engineering Education, 1994 **The Electrical Review**, 1902 **Government Reports Announcements & Index**, 1990 Metals Abstracts, 1992 **Science** John Michels (Journalist), 2010 Theoretical Chemical Engineering Abstracts, 1976 **Camera**, 1972

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