

LIMITING REACTANT STOICHIOMETRY WORKSHEET

Name _____	Section _____	Date _____
Limiting Reactants Directions: Determine the limiting reactant in each problem.		
1) In $2NaOH + MgCl_2 \rightarrow Mg(OH)_2 + 2NaCl$, 0.0100 grams of NaOH react with 2.712 grams of MgCl ₂ .		
2) 0.0718 grams of HCl react with 0.0000 grams of Ca(OH) ₂ in the reaction $2HCl + Ca(OH)_2 \rightarrow 2H_2O + CaCl_2$.		
3) 0.0718 grams of HCl react with 0.0000 grams of Ca(OH) ₂ in the reaction $2HCl + Ca(OH)_2 \rightarrow 2H_2O + CaCl_2$.		
4) 0.0000 grams of nitrogen in the reaction $N_2 + 3H_2 \rightarrow 2 NH_3$ combine with 0.0000 grams of hydrogen.		
5) In the reaction $2C_2H_2 + 5O_2 \rightarrow 4CO_2 + 2H_2O$, 1.0000 grams of O ₂ react with 0.000 grams of C ₂ H ₂ .		
6) In the reaction $2Al(OH)_3 + 3H_2SO_4 \rightarrow Al_2(SO_4)_3 + 6H_2O$, 1.38 grams of Al(OH) ₃ and 1.012 grams of H ₂ SO ₄ combine.		
7) 0.000 grams of sodium and 0.7000 grams of sodium nitrate react in the following process: $2Na(s) + 2NaNO_3 \rightarrow 2Na_2O + N_2$.		
8) 0.2000 grams of Al ₂ O ₃ react with 1.212 grams of K ₂ SO ₄ to produce products in the following reaction: $Al_2O_3 + 3K_2SO_4 \rightarrow 3Al_2(SO_4)_3 + 2K_2O$.		
9) When K ₂ PO ₄ and NaOH react in $K_2PO_4 + 3NaOH \rightarrow 2NaPO_4 + Na_2PO_4$, 0.43 grams of NaOH react with 0.88 grams K ₂ PO ₄ .		
10) 0.4000 grams of H ₂ SO ₄ combine with 0.3400 grams of Ba(OH) ₂ in the reaction $2H_2SO_4 + Ba(OH)_2 \rightarrow Ba(SO_4)_2 + 2H_2O$.		

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