

Gas Laws

21. Q: What is the relationship between volume and moles in Avogadro's Principle?

A: Volume is directly proportional to the number of moles.

22. Q: Calculate the pressure of 1.0 mole of gas in a 10.0 L container at 273 K using the ideal gas law.

A: The pressure is 2.24 atm.

23. Q: According to the ideal gas law, what happens to the pressure if the volume is reduced by half while the temperature and number of moles remain constant?

A: The pressure doubles.

24. Q: If the volume of a gas is tripled and the temperature is doubled, what happens to the pressure according to the combined gas law?

A: The pressure decreases to $\frac{2}{3}$ of its original value.

25. Q: Using gas stoichiometry, how many liters of CO_2 are produced from 5.0 L of O_2 in the reaction $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ at STP?

A: 5.0 L of CO_2 are produced.

26. Q: In the reaction $2\text{NO} + \text{O}_2 \rightarrow 2\text{NO}_2$, how many liters of NO_2 are produced from 4.0 L of NO at STP?

A: 4.0 L of NO_2 are produced.

27. Q: Calculate the volume of N_2 gas required to produce 20.0 L of NH_3 in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ at STP.

A: 10.0 L of N_2 are required.

28. Q: If the temperature of a gas is increased from 300 K to 600 K and the volume is kept constant, what happens to the pressure?

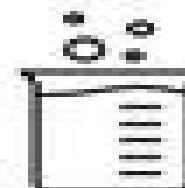
A: The pressure doubles.

29. Q: Using the ideal gas law, how do you calculate the volume of gas if you know the pressure, temperature, and number of moles?

A: $V = nRT/P$

30. Q: In the reaction $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$, how many liters of HCl are produced from 3.0 L of H_2 at STP?

A: 6.0 L of HCl are produced.



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change [An Introductory Guide to EC Competition Law and Practice](#) Valentine Korah, 1994

Guide to the Current Periodicals and Serials of the United States and Canada , 1920

[Risk Allocation and Distributive Justice in the Energy Industry](#) Smith I Azubuike, 2024-09-30

Brings the concept of gross negligence to the fore and highlights how distributive justice forms a better foundation for risk allocation in the offshore energy industry. Assessed the practice of risk allocation in gross negligence cases in offshore petroleum drilling contracts. Presented a public policy perspective on risk allocation in offshore drilling contracts. Discussed gross negligence as a *sui generis* risk and provides a definitional pathway for determining when gross negligence has occurred and how it should apply to offshore energy drilling contracts to encourage a pollution free drilling operation.

Advanced the concept of distributive justice as a basis for risk allocation between participants when a downside arises due to gross negligence. This book examines the practice of risk allocation in the offshore energy industry through the public policy lens and offers a novel perspective on the concept of gross negligence in risk allocation. This perspective is founded on the proportionality element of distributive justice in burden distribution. The assessment of how mutual indemnity clauses apply as an absolute shield against liability arising from gross negligence reveals that moral hazard can result from the practice. In the analysis this book considers the risk allocation practice in PSC and Concession regimes and how parties liability is determined in drilling contracts. This book considers gross negligence a *sui generis* risk and provides a definitional pathway for determining when gross negligence occurs and how it should apply to offshore drilling contracts. Thus it advances an environmental sustainability approach to offshore petroleum drilling operations. This book will be useful to operators and contractors resource rich countries insurance companies practitioners scholars and academics interested in risk allocation in

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