

Binary Ionic Compounds:

- Consist of only _____ elements.
- Name the _____ ion (the _____).
- Name the _____ ion (the _____), changing the ending to _____.
- When metals that can form _____ than one type of ion, in a compound, use a _____ in parentheses after the name of the _____ to show the _____.

Examples: NaCl _____ MgO _____ Ca_3S_2 _____ SrCl_2 _____**Ternary Ionic Compounds:**

- Made up of _____ elements.
- Name the _____ then name the _____ without changing the ending to "ide".

Examples: Na_2SO_4 _____ Fe_2O_3 _____

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