

Binary Ionic Compounds

- Consist of only _____ elements.
- Name the _____ ion (the _____).
- Name the _____ ion (the _____), changing the ending to _____.
- When metals that can form _____ than one type of _____ are in a compound, use a _____ in parentheses after the name of the _____ to show the _____.

Examples: NaCl _____

MgO _____

Ca_2S _____

SnCl_4 _____

Ternary Ionic Compounds

- Made up of _____ elements.
- Name the _____ then name the _____ without changing the ending to "ide".

Examples: Na_2SO_4 _____

FeCl_2O_4 _____

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N Noddings



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