

5. For the equilibrium, $A(g) \rightleftharpoons B(g)$, ΔH is -40 kJ/mol . If the ratio of the activation energies of the forward (E_f) and reverse (E_b) reactions is $\frac{2}{3}$, then :

- (1) $E_f = 60 \text{ kJ/mol}$; $E_b = 100 \text{ kJ/mol}$
- (2) $E_f = 30 \text{ kJ/mol}$; $E_b = 70 \text{ kJ/mol}$
- (3) $E_f = 80 \text{ kJ/mol}$; $E_b = 120 \text{ kJ/mol}$
- (4) $E_f = 70 \text{ kJ/mol}$; $E_b = 30 \text{ kJ/mol}$

6. Determination of the molar mass of acetic acid in benzene using freezing point depression is affected by :

- (1) dissociation
- (2) association
- (3) partial ionization
- (4) complex formation

7. The increase of pressure on ice-water system at constant temperature will lead to :

- (1) no effect on the equilibrium
- (2) a decrease in the entropy of the system
- (3) a shift of the equilibrium in the forward direction
- (4) an increase in the Gibbs energy of the system

8. At 298 K, the standard reduction potentials are 1.51 V for $\text{MnO}_4^- \text{Mn}^{2+}$, 1.36 V for $\text{Cl}_2|\text{Cl}^-$, 1.07 V for $\text{Br}_2|\text{Br}^-$, and 0.54 V for $\text{I}_2|\text{I}^-$. At pH = 3, permanganate is expected to oxidize : $\frac{RT}{F} = 0.059 \text{ V}$

- (1) Cl^- , Br^- and I^-
- (2) Cl^- and Br^-
- (3) Br^- and I^-
- (4) I^- only

5. इस साम्यवस्था के लिए, $A(g) \rightleftharpoons B(g)$, $\Delta H = -40 \text{ kJ/mol}$ है। अगर अग्र (E_f) और पश्चात्यावधी (E_b) अधिक्रियाओं के सक्रियता क्रांतियों का अनुपात $\frac{2}{3}$ है, तब :

- (1) $E_f = 60 \text{ kJ/mol}$; $E_b = 100 \text{ kJ/mol}$
- (2) $E_f = 30 \text{ kJ/mol}$; $E_b = 70 \text{ kJ/mol}$
- (3) $E_f = 80 \text{ kJ/mol}$; $E_b = 120 \text{ kJ/mol}$
- (4) $E_f = 70 \text{ kJ/mol}$; $E_b = 30 \text{ kJ/mol}$

6. ऐसिटिक अम्ल का सोलर द्रव्यमान विद्युत में हिमांक अवरोधन से निकाले जाने पर जिससे प्रभावित होता है यह है :

- (1) विद्योजन
- (2) संगृहण
- (3) अस्थिर आवर्तीकरण
- (4) संकुल विस्तरण

7. विद्युत ताप पर, अधिक दाप उत्पन्ने पर यह— जल समुद्रतल में :

- (1) साम्यवस्था पर कोई प्रभाव नहीं होता है।
- (2) निकाय की ऐन्ट्रॉपी घटती है।
- (3) साम्यवस्था अप दिशा में अस्थिर होती है।
- (4) निकाय की निष्पत्ति घटती है।

8. 298 K पर मानक अपचयन विभव, $\text{MnO}_4^- \text{Mn}^{2+}$ का 1.51 V, $\text{Cl}_2|\text{Cl}^-$ का 1.36 V, $\text{Br}_2|\text{Br}^-$ का 1.07 V और $\text{I}_2|\text{I}^-$ का 0.54 V है। pH = 3 पर MnO_4^- किसका उपचयन करेगा ?

$$\frac{RT}{F} = 0.059 \text{ V}$$

- (1) Cl^- , Br^- और I^-
- (2) Cl^- और Br^-
- (3) Br^- और I^-
- (4) I^- भाव

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Table of Contents Jee Question Papers

1. Understanding the eBook Jee Question Papers
 - The Rise of Digital Reading Jee Question Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Jee Question Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jee Question Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jee Question Papers
 - Personalized Recommendations
 - Jee Question Papers User Reviews and Ratings
 - Jee Question Papers and Bestseller Lists
5. Accessing Jee Question Papers Free and Paid eBooks

- Jee Question Papers Public Domain eBooks
- Jee Question Papers eBook Subscription Services
- Jee Question Papers Budget-Friendly Options

6. Navigating Jee Question Papers eBook Formats

- ePUB, PDF, MOBI, and More
- Jee Question Papers Compatibility with Devices
- Jee Question Papers Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Jee Question Papers
- Highlighting and Note-Taking Jee Question Papers
- Interactive Elements Jee Question Papers

8. Staying Engaged with Jee Question Papers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Jee Question Papers

9. Balancing eBooks and Physical Books Jee Question Papers

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Jee Question Papers

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Jee Question Papers

- Setting Reading Goals Jee Question Papers
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Jee Question Papers

- Fact-Checking eBook Content of Jee Question Papers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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