

5. For the equilibrium, $A(g) \rightleftharpoons B(g)$, ΔH is -40 kJ/mol . If the ratio of the activation energies of the forward (E_f) and reverse

(E_b) reactions is $\frac{2}{3}$ then :

- (1) $E_f = 60 \text{ kJ/mol}$; $E_b = 100 \text{ kJ/mol}$
- (2) $E_f = 30 \text{ kJ/mol}$; $E_b = 70 \text{ kJ/mol}$
- (3) $E_f = 80 \text{ kJ/mol}$; $E_b = 120 \text{ kJ/mol}$
- (4) $E_f = 70 \text{ kJ/mol}$; $E_b = 30 \text{ kJ/mol}$

6. Determination of the molar mass of acetic acid in benzene using freezing point depression is affected by :

- (1) dissociation
- (2) association
- (3) partial ionization
- (4) complex formation

7. The increase of pressure on ice = water system at constant temperature will lead to :

- (1) no effect on the equilibrium
- (2) a decrease in the entropy of the system
- (3) a shift of the equilibrium in the forward direction
- (4) an increase in the Gibbs energy of the system

8. At 298 K, the standard reduction potentials are 1.51 V for $\text{MnO}_4^-/\text{Mn}^{2+}$, 1.36 V for Cl_2/Cl^- , 1.07 V for Br_2/Br^- , and 0.54 V for I_2/I^- . At pH = 3, permanganate is expected to oxidize :

$$\frac{RT}{F} = 0.059 \text{ V}$$

- (1) Cl^- , Br^- and I^-
- (2) Cl^- and Br^-
- (3) Br^- and I^-
- (4) I^- only

5. इस साम्यावस्था के लिए, $A(g) \rightleftharpoons B(g)$, ΔH -40 kJ/mol है। अगर अग्र (E_f) और पश्चगामी (E_b) अभिक्रियाओं के सक्रियण ऊर्जाओं का अनुपात

$\frac{2}{3}$ है, तब :

- (1) $E_f = 60 \text{ kJ/mol}$; $E_b = 100 \text{ kJ/mol}$
- (2) $E_f = 30 \text{ kJ/mol}$; $E_b = 70 \text{ kJ/mol}$
- (3) $E_f = 80 \text{ kJ/mol}$; $E_b = 120 \text{ kJ/mol}$
- (4) $E_f = 70 \text{ kJ/mol}$; $E_b = 30 \text{ kJ/mol}$

6. ऐसिटिक अम्ल का मोलर द्रव्यमान बेंजीन में हिमांक अवनमन से निकाले जाने पर जिससे प्रभावित होता है वह है :

- (1) वियोजन
- (2) संगुणन
- (3) आंशिक आयनीकरण
- (4) संकुल बिरचन

7. स्थिर ताप पर, अधिक दाब डालने पर बर्फ = जल समुदाय में :

- (1) साम्यावस्था पर कोई प्रभाव नहीं होता है।
- (2) निकाय की ऐन्ट्रॉपी घटती है।
- (3) साम्यावस्था अग्र दिशा में अग्रसर होती है।
- (4) निकाय की गिब्स ऊर्जा बढ़ती है।

8. 298 K पर मानक अपचयन विभव, $\text{MnO}_4^-/\text{Mn}^{2+}$ का 1.51 V, Cl_2/Cl^- का 1.36 V, Br_2/Br^- का 1.07 V और I_2/I^- का 0.54 V है। pH = 3 पर MnO_4^- किसका उपचयन करेगा ?

$$\frac{RT}{F} = 0.059 \text{ V}$$

- (1) Cl^- , Br^- और I^-
- (2) Cl^- और Br^-
- (3) Br^- और I^-
- (4) I^- मात्र

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TABLE OF CONTENT

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TABLE OF CONTENT

Solved Papers 2006
 2019 JEE Main Advanced
 2019 JEE Main Advanced
 2017 JEE Main Advanced
 2016 JEE Main Advanced
 2015 JEE Main Advanced
 2014 JEE Main Advanced
 2013 IIT JEE
 2012 IIT JEE
 2011 IIT JEE
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 2009 IIT JEE
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 - The Rise of Digital Reading Jee Question Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Jee Question Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jee Question Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jee Question Papers
 - Personalized Recommendations
 - Jee Question Papers User Reviews and Ratings
 - Jee Question Papers and Bestseller Lists
5. Accessing Jee Question Papers Free and Paid eBooks

- Jee Question Papers Public Domain eBooks
- Jee Question Papers eBook Subscription Services
- Jee Question Papers Budget-Friendly Options
- 6. Navigating Jee Question Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Jee Question Papers Compatibility with Devices
 - Jee Question Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jee Question Papers
 - Highlighting and Note-Taking Jee Question Papers
 - Interactive Elements Jee Question Papers
- 8. Staying Engaged with Jee Question Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jee Question Papers
- 9. Balancing eBooks and Physical Books Jee Question Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jee Question Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Jee Question Papers
 - Setting Reading Goals Jee Question Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jee Question Papers
 - Fact-Checking eBook Content of Jee Question Papers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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