

Name the following acids

1. HCL hydrochloric acid
2. HClO<sub>4</sub> perchloric acid
3. HIO<sub>3</sub> iodic acid
4. HI hydroiodic acid
5. H<sub>2</sub>SO<sub>4</sub> sulfuric acid
6. H<sub>2</sub>S hydrosulfuric acid
7. HCN hydrocyanic acid
8. H<sub>2</sub>CO<sub>3</sub> carbonic acid
9. HBrO<sub>4</sub> perbromic acid
10. HBrO<sub>3</sub> bromic acid
11. HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> acetic acid
12. H<sub>3</sub>PO<sub>3</sub> phosphorous acid
13. H<sub>3</sub>P hydrophosphoric acid
14. H<sub>2</sub>CrO<sub>4</sub> chromic acid
15. H<sub>2</sub>CrO<sub>2</sub> hypochromous acid
16. H<sub>2</sub>Cr hydrochromic acid

Give formulas for the following acids

17. hydrofluoric acid HF
18. nitric acid HNO<sub>3</sub>
19. nitrous acid HNO<sub>2</sub>
20. hydroxic acid H<sub>2</sub>O
21. periodic acid HIO<sub>4</sub>
22. percarbonic acid H<sub>2</sub>CO<sub>4</sub>
23. hypobromous acid HBrO
24. bromous acid HBrO<sub>2</sub>
25. permanganic acid HMnO<sub>4</sub>
26. manganic acid HMnO<sub>3</sub>
27. iodous acid HIO<sub>2</sub>
28. sulfurous acid H<sub>2</sub>SO<sub>3</sub>
29. sulfuric acid H<sub>2</sub>SO<sub>4</sub>
30. perchloric acid HClO<sub>4</sub>
31. chlorous acid HClO<sub>2</sub>
32. acetic acid HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>
33. phosphoric acid H<sub>3</sub>PO<sub>4</sub>
34. phosphorous acid H<sub>3</sub>PO<sub>3</sub>

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**Mike Jess**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the name bar.

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