



- 1) What is happening to the average kinetic energy of the molecules in the sample during section 2? **remains constant**
- 2) As a substance goes through section (2), what happens to the distance between the particles? **increases**
- 3) What is the name of the process happening during section (4)? **boiling (vaporization)**
- 4) What would be the name of the process happening during section (4) if time were going the other way? **condensation**
- 5) What is the melting point of this substance? **0 °C**
- 6) At what temperature would this sample finish boiling? **100 °C**
- 7) When this substance is melting, the temperature of the ice-water mixture remains constant because:
 - d. **Heat energy is being converted to kinetic energy**
- 8) When a given quantity of water is heated at a constant rate, the phase change from liquid to gas takes longer than the phase change from solid to liquid because
 - a. **The heat of vaporization is greater than the heat of fusion**
- 9) The temperature at which a substance in the liquid state freezes is the same as the temperature at which the substance
 - a. **Melts**
- 10) Is this curve showing an **endothermic** or an exothermic process?
- 11) 34°C is equal to **307 K**.
- 12) -128°C is equal to **145 K**.

Heating Curve Problems Key

L Cohen

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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General Chemistry Key Concise Review Guide Sixth Edition Darrell D. Ebbing, 1999 Catalog of National Bureau of Standards Publications, 1966-1976: Citations and abstracts United States. National Bureau of Standards. Technical Information and Publications Division, 1978 **Publications of the National Bureau of Standards ... Catalog** United States. National Bureau of Standards, 1975 **Publications of the National Institute of Standards and Technology ... Catalog** National Institute of Standards and Technology (U.S.), 1976 **NBS Special Publication** , 1968 **Catalog of National Bureau of Standards Publications, 1966-1976** United States. National Bureau of Standards, 1978

Publications United States. National Bureau of Standards, 1976 Radiant Floor Heating, Second Edition R. Dodge Woodson, 2010-01-04 The latest radiant floor heating methods and materials In this fully updated guide master plumber and long time contractor R Dodge Woodson explains step by step how to install test and debug in floor heating systems the healthy quiet economical and clean heating method Radiant Floor Heating Second Edition now offers details on alternative fuel sources and covers new material options and installation procedures updated code information and state of the art piping and heating equipment Revised illustrations featuring modern products are included in this practical resource Get everything you need to Understand and explain the advantages of in floor radiant heating Select superior equipment tubing boilers expansion tanks pumps and controls Install embedded and dry radiant heat systems in new or existing homes Incorporate hot water heating into your designs Set up heating zones for variable temperatures throughout a house Design a solar powered system Use alternative fuel sources Provide outdoor ice removal and other amenities Troubleshoot and repair system problems *Notes on the Plotting of Speed-time Curves* Cyprien Odilon Mailloux, 1902 **The Engineer** , 1915

Forging, Stamping, Heat Treating , 1929 **English Mechanic and Mirror of Science and Art** , 1875 **Machine Design Drawing Room Problems** Calvin Dodge Albert, 1923 Railway Mechanical Engineer , 1921 **Combustion** , 1923 *Engineering and Finance* , 1926 **American Railroad Journal** , 1921 **English Mechanic and World of Science** , 1873 **Scientific American** , 1915 Power , 1929

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