

Limiting Reactant Worksheet

- 1 Balance the following reaction:



- a Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

- b How many grams of excess reactant are left?

- 2 Acrylonitrile ($\text{C}_3\text{H}_3\text{N}$) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene (C_3H_6) by reacting with nitric oxide (NO) as follows:



What mass of $\text{C}_3\text{H}_3\text{N}$ can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

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Limiting Reactant and Percent Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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