

$$\text{Rate}_1 = \frac{0.0001 \text{ g} - 0.00001 \text{ g}}{0.0001 \text{ mol}} = \frac{999.99 \text{ g}}{0.0001 \text{ mol}} = 9.9999 \times 10^6 \text{ g/mol}$$

4. Calculate that the molarity is the same molarity that you calculated in step 3.

Yes, the molarity is the same as the molarity calculated in step 3.

#### Part B - Unknown solutions

Using a method similar to the one used in part A, calculate and carry out a procedure for determining the molarities of the unknown Ag<sup>+</sup>(aq) solutions and the unknown Pb<sup>2+</sup>(aq) solutions. Present your data and show your calculations.

#### Unknown Ag<sup>+</sup>(aq)

Open the solution of which Pb<sup>2+</sup>, and Ag<sup>+</sup>(aq) was found.

Ag<sup>+</sup>(aq) found (g) = 0.27000000000000002

$$\text{Rate}_2 = \frac{0.0001 \text{ g} - 0.27000000000000002 \text{ g}}{0.0001 \text{ mol}} = \frac{-0.26999999999999998 \text{ g}}{0.0001 \text{ mol}} = -2.6999999999999998 \times 10^6 \text{ g/mol}$$

$$\text{Rate}_3 = \frac{0.0001 \text{ g} - 0.27000000000000002 \text{ g}}{0.0001 \text{ mol}} = \frac{-0.26999999999999998 \text{ g}}{0.0001 \text{ mol}} = -2.6999999999999998 \times 10^6 \text{ g/mol}$$

#### Unknown Pb<sup>2+</sup>(aq)

Open the solution of which Ag<sup>+</sup>(aq), and Pb<sup>2+</sup>(aq) was found.

Pb<sup>2+</sup>(aq) found (g) = 0.00010000000000000002

$$\text{Rate}_4 = \frac{0.0001 \text{ g} - 0.00010000000000000002 \text{ g}}{0.0001 \text{ mol}} = \frac{-0.00009999999999999998 \text{ g}}{0.0001 \text{ mol}} = -9.999999999999998 \times 10^3 \text{ g/mol}$$

$$\text{Rate}_5 = \frac{0.0001 \text{ g} - 0.00010000000000000002 \text{ g}}{0.0001 \text{ mol}} = \frac{-0.00009999999999999998 \text{ g}}{0.0001 \text{ mol}} = -9.999999999999998 \times 10^3 \text{ g/mol}$$

#### Conclusion

5. If you have a company that is using this method to remove Ag<sup>+</sup>(aq), knowing that Ag<sup>+</sup>(aq) is an essential (when do you think) would the a better cleaning method to use, Ag<sup>+</sup>(aq) or Pb<sup>2+</sup>(aq)?

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**American Chemical Society**

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*The Organic Chem Lab Survival Manual* James W. Zubrick, 2016-01-19 Written for the laboratory that accompanies the sophomore junior level courses in Organic Chemistry Zubrick provides students with a valuable guide to the basic techniques of the Organic Chemistry lab The book will help students understand and practice good lab safety It will also help students become familiar with basic instrumentation techniques and apparatus and help them master the latest techniques such as interpretation of infrared spectroscopy The guide is mostly macroscale in its orientation

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*Fundamentals of Chemical Reaction Engineering* Mark E. Davis, Robert J. Davis, 2013-05-27 Appropriate for a one semester undergraduate or first year graduate course this text introduces the quantitative treatment of chemical reaction engineering It covers both homogeneous and heterogeneous reacting systems and examines chemical reaction engineering as well as chemical reactor engineering Each chapter contains numerous worked out problems and real world vignettes involving commercial applications a feature widely praised by reviewers and teachers 2003 edition

*General Chemistry Key Concise Review Guide Sixth Edition* Darrell D. Ebbing, 1999 *Chemical and Catalytic Reaction Engineering* James J. Carberry, 2001-01-01 Designed to give chemical engineers background for managing chemical reactions this text examines the behavior of chemical reactions and reactors conservation equations for reactors heterogeneous reactions fluid fluid and fluid solid reaction systems heterogeneous catalysis and catalytic kinetics diffusion and heterogeneous catalysis and analyses and design of heterogeneous reactors 1976 edition

*Chemistry in the Community (Enhanced Core Four)* American Chemical Society, 2006-02-15

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**ChemCom** American Chemical Society, 1993      **Semimicro Experiments in Organic Chemistry** Bruce Nelson Campbell, Ernest Ray Kline, 1964      *SourceBook Version 2.1, 1998*      Addison-Wesley Chemistry, 1990      *Chemistry* Neil D. Jespersen, Alison Hyslop, 2021-12-03 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental

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