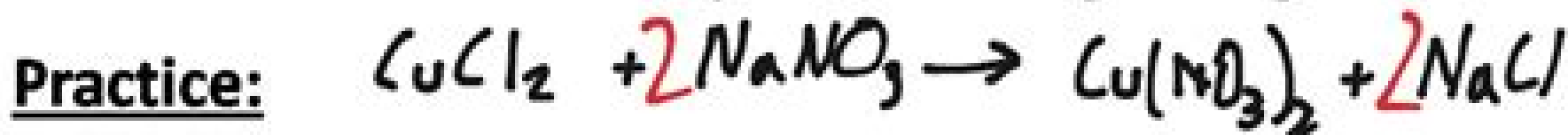


9.2 – Stoichiometry – Limiting Reagent



20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent.

$$20\text{g } \cancel{\text{CuCl}_2} \times \frac{1 \text{ mol } \cancel{\text{CuCl}_2}}{137.452\text{g } \cancel{\text{CuCl}_2}} = 0.149 \text{ mol } \cancel{\text{CuCl}_2} \times \frac{2 \text{ mol } \text{NaNO}_3}{1 \text{ mol } \cancel{\text{CuCl}_2}} = 0.298 \text{ mol } \text{NaNO}_3$$

HAVE Need

$$25\text{g } \cancel{\text{NaNO}_3} \times \frac{1 \text{ mol } \cancel{\text{NaNO}_3}}{84.994\text{g } \cancel{\text{NaNO}_3}} = 0.294 \text{ mol } \cancel{\text{NaNO}_3} \times \frac{1 \text{ mol } \text{CuCl}_2}{2 \text{ mol } \cancel{\text{NaNO}_3}} = 0.147 \text{ mol } \text{CuCl}_2$$

You need 0.298 mol NaNO_3 , but you have 0.294 mol NaNO_3

Limiting Reagent Exam Stoichiometry

M Mark



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exemplary practice supported by solid evidence and to include practitioners who offer models of teaching and learning that meet the high standards of the scientific disciplines Our intention is to let these distinguished scientists speak for themselves and to offer authentic guidance to those who seek models of excellence Our primary audience consists of the thousands of dedicated faculty and graduate students who teach undergraduate science at community and technical colleges 4 year liberal arts institutions comprehensive regional campuses and flagship research universities In keeping with Wieman s challenge our primary focus has been on identifying classroom practices that encourage and support meaningful learning and conceptual understanding in the natural sciences The content is structured as follows after an Introduction based on Constructivist Learning Theory Section I the practices we explore are Eliciting Ideas and Encouraging Reflection Section II Using Clickers to Engage Students Section III Supporting Peer Interaction through Small Group Activities Section IV Restructuring Curriculum and Instruction Section V Rethinking the Physical Environment Section VI Enhancing Understanding with Technology Section VII and Assessing Understanding Section VIII The book s final section IX is devoted to Professional Issues facing college and university faculty who choose to adopt active learning in their courses The common feature underlying all of the strategies described in this book is their emphasis on actively engaging students who seek to make sense of natural objects and events Many of the strategies we highlight emerge from a constructivist view of learning that has gained widespread acceptance in recent years In this view learners make sense of the world by forging connections between new ideas and those that are part of their existing knowledge base For most students that knowledge base is riddled with a host of na ve notions misconceptions and alternative conceptions they have acquired throughout their lives To a considerable extent the job of the teacher is to coax out these ideas to help students understand how their ideas differ from the scientifically accepted view to assist as students restructure and reconcile their newly acquired knowledge and to provide opportunities for students to evaluate what they have learned and apply it in novel circumstances Clearly this prescription demands far more than most college and university scientists have been prepared for

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conceptualised as building of tentative models that provide greater explanatory power to students understanding Journal of the Pennsylvania Academy of Science, 2001

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