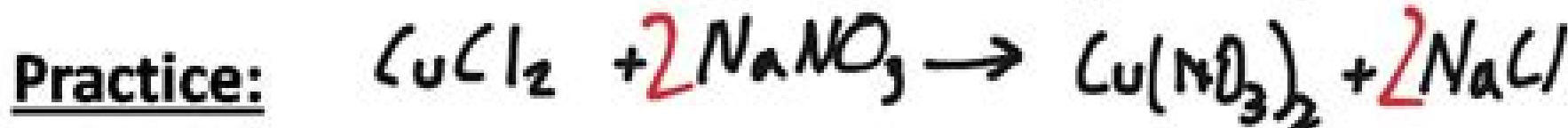


9.2 – Stoichiometry – Limiting Reagent



20 grams of copper (II) chloride react with 25 grams of sodium nitrate. Identify the limiting reagent.

$$\begin{aligned} 20 \text{ g } \text{CuCl}_2 &\times \frac{1 \text{ mol CuCl}_2}{137.452 \text{ g CuCl}_2} = 0.147 \text{ mol CuCl}_2 \quad \text{HAVE} \quad \text{Need} \\ &\times \frac{2 \text{ mol NaNO}_3}{1 \text{ mol CuCl}_2} = 0.294 \text{ mol NaNO}_3 \end{aligned}$$
$$\begin{aligned} 25 \text{ g } \text{NaNO}_3 &\times \frac{1 \text{ mol NaNO}_3}{84.994 \text{ g NaNO}_3} = 0.294 \text{ mol NaNO}_3 \times \frac{1 \text{ mol CuCl}_2}{2 \text{ mol NaNO}_3} = 0.147 \text{ mol CuCl}_2 \end{aligned}$$

You need 0.294 mol NaNO_3 , but you have 0.294 mol NaNO_3

Limiting Reagent Exam Stoichiometry

M Mark

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exemplary practice supported by solid evidence and to include practitioners who offer models of teaching and learning that meet the high standards of the scientific disciplines Our intention is to let these distinguished scientists speak for themselves and to offer authentic guidance to those who seek models of excellence Our primary audience consists of the thousands of dedicated faculty and graduate students who teach undergraduate science at community and technical colleges 4 year liberal arts institutions comprehensive regional campuses and flagship research universities In keeping with Wieman s challenge our primary focus has been on identifying classroom practices that encourage and support meaningful learning and conceptual understanding in the natural sciences The content is structured as follows after an Introduction based on Constructivist Learning Theory Section I the practices we explore are Eliciting Ideas and Encouraging Reflection Section II Using Clickers to Engage Students Section III Supporting Peer Interaction through Small Group Activities Section IV Restructuring Curriculum and Instruction Section V Rethinking the Physical Environment Section VI Enhancing Understanding with Technology Section VII and Assessing Understanding Section VIII The book s final section IX is devoted to Professional Issues facing college and university faculty who choose to adopt active learning in their courses The common feature underlying all of the strategies described in this book is their emphasis on actively engaging students who seek to make sense of natural objects and events Many of the strategies we highlight emerge from a constructivist view of learning that has gained widespread acceptance in recent years In this view learners make sense of the world by forging connections between new ideas and those that are part of their existing knowledge base For most students that knowledge base is riddled with a host of na ve notions misconceptions and alternative conceptions they have acquired throughout their lives To a considerable extent the job of the teacher is to coax out these ideas to help students understand how their ideas differ from the scientifically accepted view to assist as students restructure and reconcile their newly acquired knowledge and to provide opportunities for students to evaluate what they have learned and apply it in novel circumstances Clearly this prescription demands far more than most college and university scientists have been prepared for Teaching General Chemistry Mansoor Niaz, 2008 The main objective of this monograph is to incorporate history and philosophy of science in the chemistry curriculum in order to provide students an overview of the dynamics of scientific research which involves controversies conflicts and rivalries among scientists that is the humanising aspects of science A major thesis of this book is the parallel between the construction of knowledge by the students and the scientists In looking for this relationship it is not necessary that ontogeny recapitulate phylogeny but rather to establish that students can face similar difficulties in conceptualising problems as those faced by the scientists in the past Given the vast amount of literature on students alternative conceptions misconceptions in science it is plausible to suggest that these can be considered not as mistakes but rather as tentative models leading to greater conceptual understanding Just as scientists resist changes in the hard core of their beliefs by offering auxiliary hypotheses students may adopt similar strategies Conceptual change in science education can thus be

conceptualised as building of tentative models that provide greater explanatory power to students understanding [Journal of the Pennsylvania Academy of Science ,2001](#) **The Best Test Preparation & Review Course FE/EIT Fundamentals**

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practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry

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