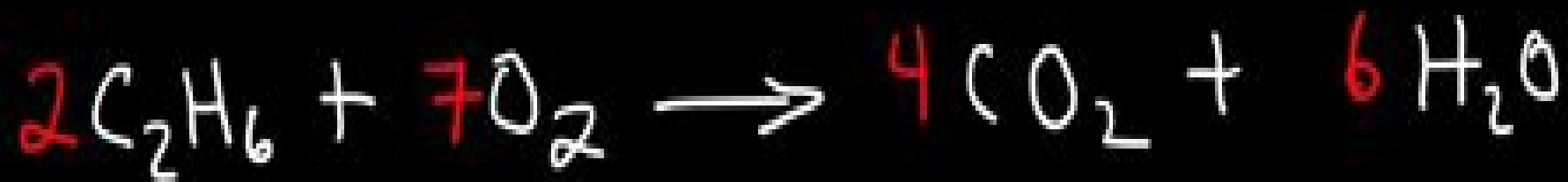


# Limiting Reactant

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$$\frac{84\text{g O}_2}{1} \times \frac{1\text{mol O}_2}{32\text{g O}_2} \times \frac{6\text{mol H}_2\text{O}}{7\text{mol O}_2} \times \frac{18.016\text{g H}_2\text{O}}{1\text{mol H}_2\text{O}}$$

$$= 40.54\text{g H}_2\text{O}$$

# Limiting Reactants C1y Vm

**Lifelique**



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