

# Limiting Reactant Worksheet

1 Balance the following reaction:



2 Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

3 How many grams of excess reactant are left?

4 Acrylonitrile ( $\text{C}_3\text{H}_3\text{N}$ ) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene ( $\text{C}_3\text{H}_6$ ) by reacting with nitric oxide (NO) as follows:



What mass of  $\text{C}_3\text{H}_3\text{N}$  can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

# Limiting Reactant Homework Answers On Chemfiesta

**Lifeliqe**

**Limiting Reactant Homework Answers On Chemfiesta:**

*Limiting Reactant and Percent* Lifeliqe, 2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

Delve into the emotional tapestry woven by Emotional Journey with in Dive into the Emotion of **Limiting Reactant Homework Answers On Chemfiesta** . This ebook, available for download in a PDF format ( Download in PDF: \*), is more than just words on a page; it is a journey of connection and profound emotion. Immerse yourself in narratives that tug at your heartstrings. Download now to experience the pulse of each page and let your emotions run wild.

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