

Quiz & Worksheet - Calculating Reaction Yield and Percentage Yield from a Limiting Reactant

1. What is theoretical yield?

- The maximum amount of product that can be made from the amount of reactants.
- The minimum amount of product that can be made from the amount of reactants.
- The measured amount of a product from a reaction.
- The measured amount of reactant in a reaction.
- The maximum amount of reactant that can be made from the amount of product.

2. What is the actual yield?

- The measured amount of a product from a reaction
- The theoretical amount of a product from a reaction
- The measured amount of a reactant in a reaction
- The theoretical amount of a reactant in a reaction
- The actual amount of a reactant

3. How is percent yield calculated?

- Percent yield = $\frac{\text{actual yield}}{\text{theoretical yield}} \times 100\%$
- Percent yield = $\frac{\text{theoretical yield}}{\text{actual yield}} \times 100\%$
- Percent yield = $\frac{\text{actual yield}}{\text{theoretical yield}}$
- Percent yield = $\frac{\text{theoretical yield}}{\text{actual yield}}$
- Percent yield = $\text{actual yield} \times \text{theoretical yield} \times 100\%$

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Mary Millhollon, Richard H. Langley

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