

# Quiz & Worksheet - Calculating Reaction Yield and Percentage Yield from a Limiting Reactant

## 1. What is theoretical yield?

- ☐ The maximum amount of product that can be made from the amount of reactants.
- ☐ The minimum amount of product that can be made from the amount of reactants.
- ☐ The measured amount of a product from a reaction.
- ☐ The measured amount of reactant in a reaction.
- ☐ The maximum amount of reactant that can be made from the amount of product.

## 2. What is the actual yield?

- ☐ The measured amount of a product from a reaction
- ☐ The theoretical amount of a product from a reaction
- ☐ The measured amount of a reactant in a reaction
- ☐ The theoretical amount of a reactant in a reaction
- ☐ The actual amount of a reactant

## 3. How is percent yield calculated?

- ☐  $\text{Percent yield} = \text{actual yield} / \text{theoretical yield} \times 100\%$
- ☐  $\text{Percent yield} = \text{theoretical yield} / \text{actual yield} \times 100\%$
- ☐  $\text{Percent yield} = \text{actual yield} / \text{theoretical yield}$
- ☐  $\text{Percent yield} = \text{theoretical yield} / \text{actual yield}$
- ☐  $\text{Percent yield} = \text{actual yield} \times \text{theoretical yield} \times 100\%$

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# Limiting Reactant And Percentage Yield Quiz

**Mary Millhollon, Richard H. Langley**



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