

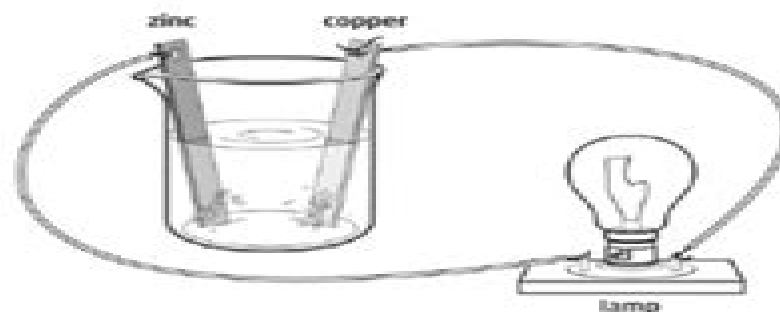
Chapter 11 Workbook

NAME: _____

Section 11.1 Review - Cells and Batteries

Multiple Choice

- Which two materials would make good electrodes for a voltaic (wet) cell?
 - copper and zinc
 - two strips of copper metal in distilled water
 - acid and copper
 - acid and zinc
 - two strips of zinc
- Which of the following is *not* an advantage of using a dry cell over a wet cell?
 - Dry cells are less easily damaged compared with wet cells.
 - Dry cells are easier to handle than voltaic cells.
 - Dry cells are used in portable radios.
 - Dry cells can be connected to make a battery, but wet cells cannot make a battery.
 - Dry cells can be transported more easily.
- What is a hybrid vehicle?
 - a vehicle that is fuelled by solar energy
 - a vehicle that is fuelled by solar energy and hydroelectric energy
 - a vehicle that is fuelled by two different types of fossil fuels
 - a vehicle that is fuelled by fossil fuels and electrical energy
 - a vehicle that is fuelled by propane only
- What types of cells can power a flashlight?
 - zinc-carbon cells
 - alkaline cells
 - rechargeable cells
 - only a voltaic wet cell
 - any of A, B, and C
- Which of the following is *true* for a voltaic cell?
 - The electrodes must be made of metal.
 - The electrolyte must be conducting.
 - The electrodes must be an insulator.
 - A and B are true.
 - B and C and true.
- Which cell has the least impact on the environment?
 - wet cell
 - dry cell
 - solar cell
 - fuel cell
 - alkaline cell
- What type of cell is a lead-acid cell?
 - primary dry cell
 - primary wet cell
 - secondary dry cell
 - secondary wet cell
 - none of the above
- Which of the following statements is *true* about the diagram below?
 - The zinc electrode forms positive ions, and electrons flow from zinc to copper through the lamp.
 - The zinc electrode forms negative ions, and electrons flow from zinc to copper through the lamp.
 - The zinc electrode forms positive ions, and electrons flow from copper to zinc through the lamp.
 - The zinc electrode forms positive ions, and protons flow from copper to zinc through the lamp.
 - The zinc electrode forms negative ions, and electrons flow from copper to zinc through the lamp.



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Roger N. Walsh, Deane H. Shapiro



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