

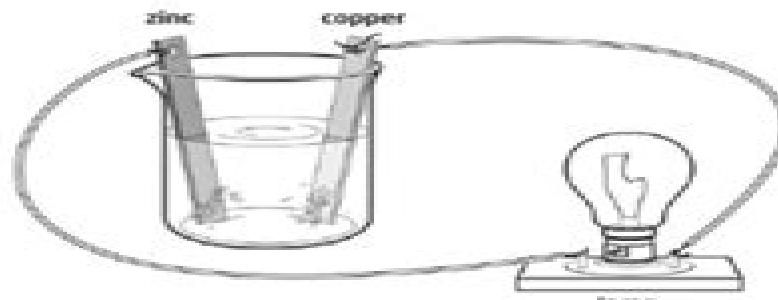
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Chapter 11 Workbook

Section 11.1 Review - Cells and Batteries

Multiple Choice

- Which two materials would make good electrodes for a voltaic (wet) cell?
 - copper and zinc
 - two strips of copper metal in distilled water
 - acid and copper
 - acid and zinc
 - two strips of zinc
- Which of the following is *not* an advantage of using a dry cell over a wet cell?
 - Dry cells are less easily damaged compared with wet cells.
 - Dry cells are easier to handle than voltaic cells.
 - Dry cells are used in portable radios.
 - Dry cells can be connected to make a battery, but wet cells cannot make a battery.
 - Dry cells can be transported more easily.
- What is a hybrid vehicle?
 - a vehicle that is fuelled by solar energy
 - a vehicle that is fuelled by solar energy and hydroelectric energy
 - a vehicle that is fuelled by two different types of fossil fuels
 - a vehicle that is fuelled by fossil fuels and electrical energy
 - a vehicle that is fuelled by propane only
- What types of cells can power a flashlight?
 - zinc-carbon cells
 - alkaline cells
 - rechargeable cells
 - only a voltaic wet cell
 - any of A, B, and C
- Which of the following is *true* for a voltaic cell?
 - The electrodes must be made of metal.
 - The electrolyte must be conducting.
 - The electrodes must be an insulator.
 - A and B are true.
 - B and C are true.
- Which cell has the least impact on the environment?
 - wet cell
 - dry cell
 - solar cell
 - fuel cell
 - alkaline cell
- What type of cell is a lead-acid cell?
 - primary dry cell
 - primary wet cell
 - secondary dry cell
 - secondary wet cell
 - none of the above
- Which of the following statements is *true* about the diagram below?



- The zinc electrode forms positive ions, and electrons flow from zinc to copper through the lamp.
- The zinc electrode forms negative ions, and electrons flow from zinc to copper through the lamp.
- The zinc electrode forms positive ions, and electrons flow from copper to zinc through the lamp.
- The zinc electrode forms positive ions, and protons flow from copper to zinc through the lamp.
- The zinc electrode forms negative ions, and electrons flow from copper to zinc through the lamp.

Interactive Science Workbook Ch11

Roger N. Walsh, Deane H. Shapiro

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Ravid, Elizabeth Oyer, 2010-04-16 The fourth edition of the Study Guide for Practical Statistics for Educators serves as a user friendly and accessible way for students to better understand, review, and apply the concepts introduced in its companion textbook, Practical Statistics for Educators (Ravid, 2011). Since the first edition of this study guide came out in 1994, thousands of students in educational statistics courses and their professors have found it to be an excellent guide with clear and easy to follow instructions and examples. The study guide allows students to reinforce and test their knowledge of the concepts addressed in each chapter of the textbook. At the end of each chapter, the best answer for each exercise is given along with an explanation for why the correct answer is better than the other choices. New in this edition are accompanying Excel exercises so students may perform data analysis with this commonly used software using data available on the web based portal that accompanies the guide.

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