

Problem: Reactions with Limiting Amounts of Reactants

Lithium oxide is used aboard the space shuttle to remove water from the air supply according to the equation:



If 80.0 g of water are to be removed and 65.0 g of Li_2O are available, which reactant is limiting? How many grams of excess reactant remain? How many grams of LiOH are produced?

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Mordechay Schlesinger,2009-08-15 The present volume is the second in a two volume set dealing with modelling and numerical simulations in electrochemistry Emphasis is placed on the aspect of nanoelectrochemical issues It seems appropriate at this juncture to mention the n growing body of opinion in some circles that George Box was right when he stated three decades ago that All models are wrong but some are useful Actually when the statement itself was made it would have been more appropriate to say that All models are inaccurate but most are useful nonetheless At present however the statement as it was made is far more appropriate and closer to the facts than ever before Currently we are in the midst of the age of massively abundant data Today s philosophy seems to be that we do not need to know why one piece of information is better than another except through the statistics of incoming and outgoing links between information and this is good enough It is why both in principle and in practice one can translate between two languages without knowledge of either While none of this can be ignored and it may even be true that All models are wrong and increasingly you can succeed without them the traditional approach of scienti c modelling is still the order of the day That approach may be stated as hypothesize measure model test It is in this light that the present volume should be viewed

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